

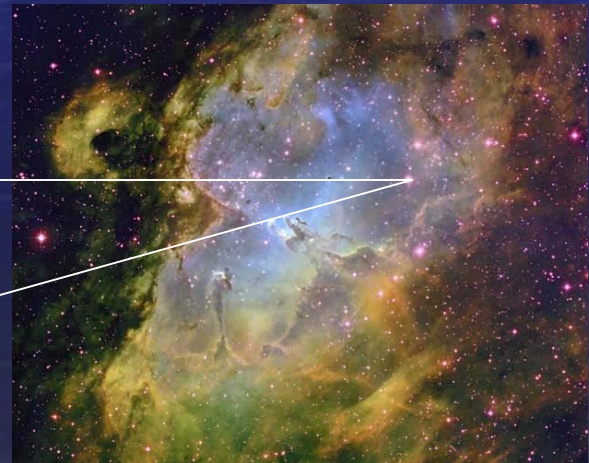
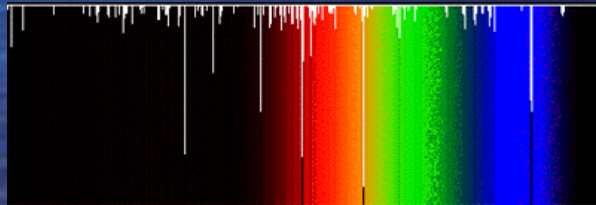
# Towards a Complete Electronic Database of PAHs and the Identification of Resolved DIBs

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# The Diffuse Interstellar Bands (DIBs)

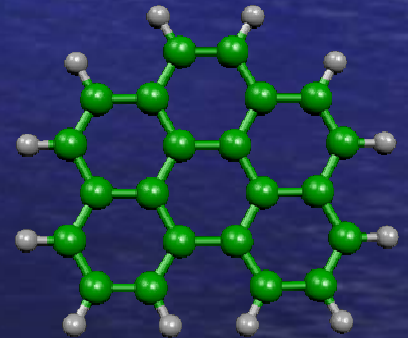
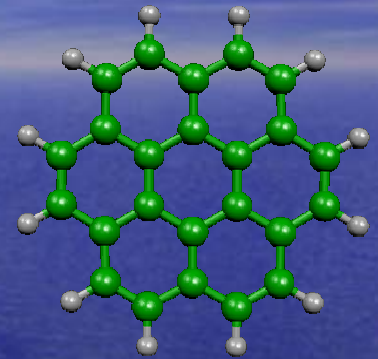
- Interstellar absorption bands first discovered by Heger in 1921. More than 300 diffuse bands in the visible and near infrared.
- No carriers have ever been identified!
- Providing crucial information on the chemical compositions of galaxies and interstellar medium, on the energy balance, chemical evolution of the universe and origin of life...





# The PAH–DIB Hypothesis

- PAHs (neutral or ionic) may be responsible for some of the DIBs.
- Abundance of elements in the universe: H, He, O, C, N, Ne...
- PAHs are remarkably photo stable.
- PAHs have been identified in meteorites.
- PAHs and the infrared emission bands.



# We Are Lottery Players



- Scientists are lottery players who are hoping to hit the Jack Pot by chance.
- Unfortunately, up to date only very limited PAHs have been studied either experimentally or theoretically:  $\sim 0.1\%$  PAHs containing up to 10 fused benzene rings.
- A big question to ask therefore is: “exactly how many PAHs are out there and what PAHs should we study?”



# Computer Enumeration of PAHs

- The ideal carbon skeleton of a PAH is a “polyhex” that consists of  $h$  fused benzene rings.
- By “ideal”, we mean all benzene rings in the polyhex are identical regular hexagons.
- How many polyhexes exist for a given number of  $h$  hexagons? Harary offered \$100 for the solution to this difficult problem in 1968.

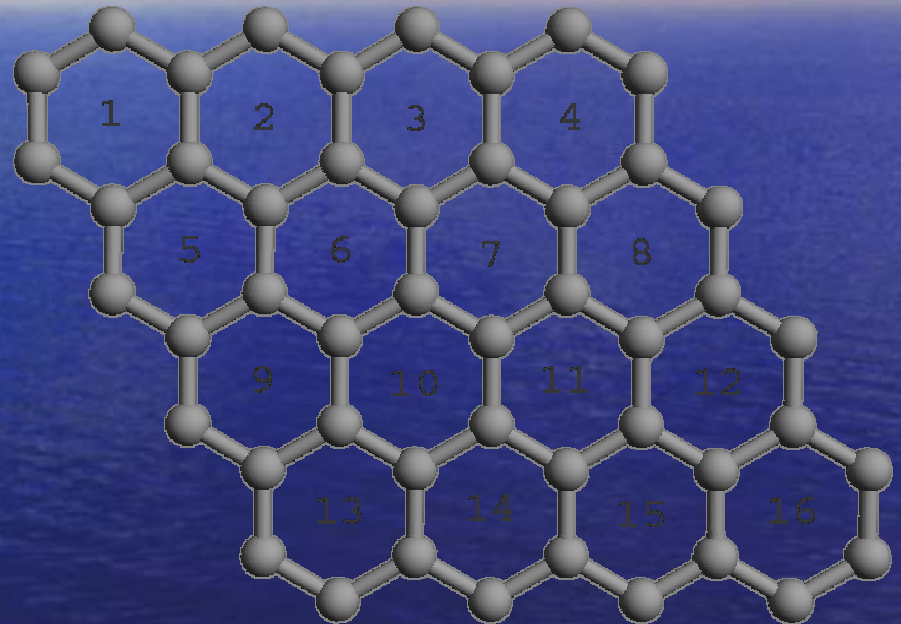
# A Working Algorithm

- Cell growth: polyhexes with  $h$  hexagons are generated from polyhexes with  $h - 1$  hexagons.
- Each time a new polyhex is generated, we transform it using each of the 12 symmetry operations.
- Each transformed polyhex is then converted into its SIR and compared to existing SIRs – the most time consuming step.
- A balanced search tree is used to store the SIRs.



# The Honeycomb Grid

- All polyhexes with  $h$  hexagons can be contained in a  $h \times h$  block honeycomb grid.
- Each hexagon, or “cell”, is represented by one grid point located in its center of mass.



A  $4 \times 4$  honeycomb grid.

# The Index Representation

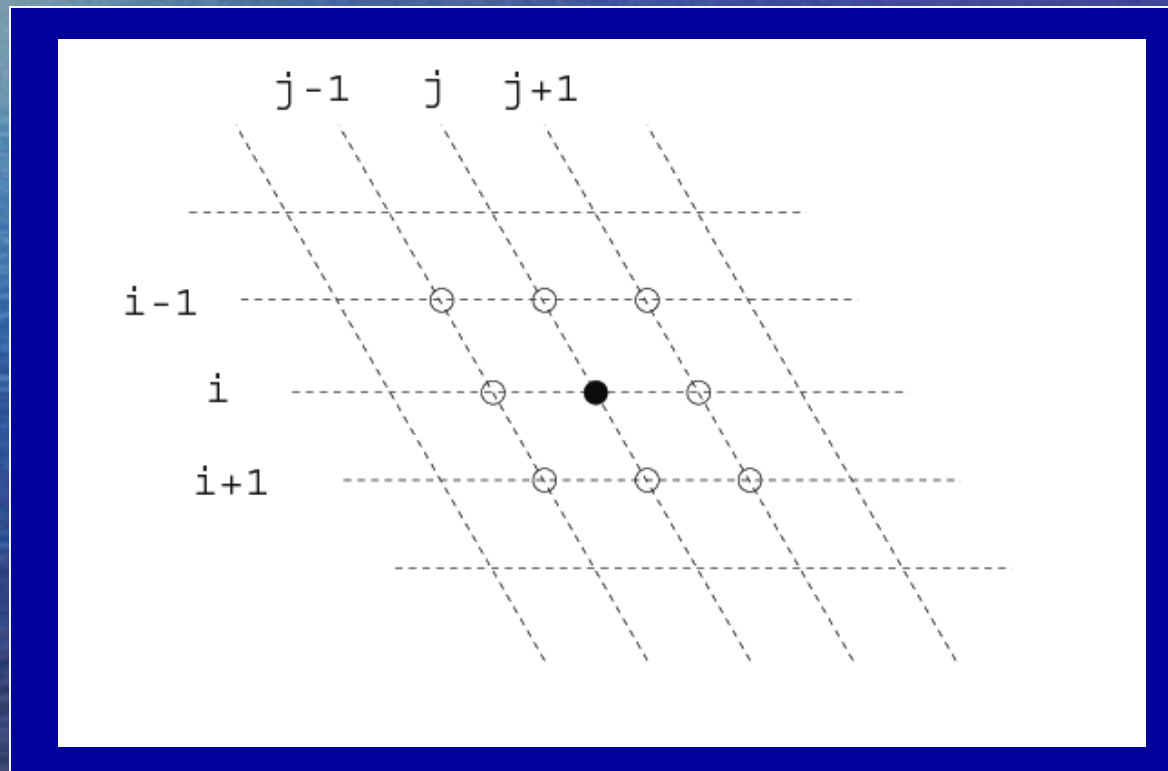
- A polyhex can be represented by the cell numbers of its constitution cells, denoted as  $C(i, j, \dots)$ , where  $i, j, \dots$  are the cell numbers.
- Index Representation: the cells of the T-shape are represented by their indices.
- Standard Index Representation (SIR).

$$C(1,5,2,6) \equiv \begin{pmatrix} 1 & 2 & 1 & 2 \\ 1 & 1 & 2 & 2 \end{pmatrix} \equiv \begin{pmatrix} 1 & 1 & 2 & 2 \\ 1 & 2 & 1 & 2 \end{pmatrix}$$



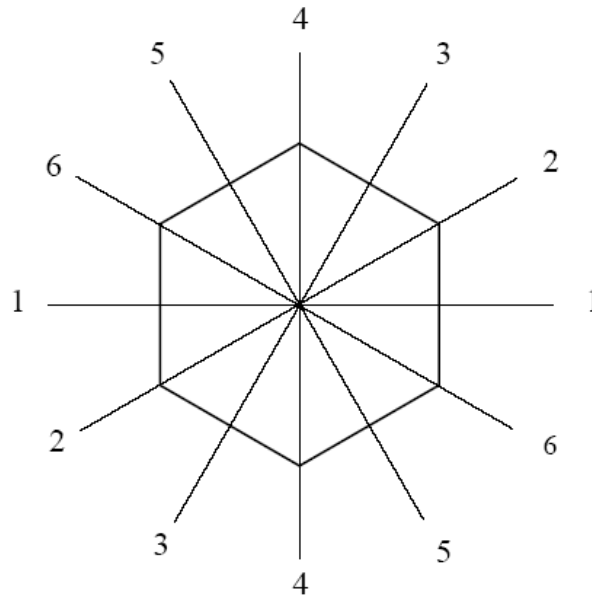
# Cell Connectivity

- Each cell (grid point) is connected with six cells (grid points):



# Symmetry Operations

- One translation.
- Six rotations:  $E$ ,  $C_6$ ,  $C_6^{-1}$ ,  $C_3$ ,  $C_3^{-1}$ ,  $C_2$ .
- Six reflections:



The six mirrors of the honeycomb grid:  $1 - \sigma_x$ ,  $2 - \sigma_y''$ ,  $3 - \sigma_x'$ ,  $4 - \sigma_y$ ,  $5 - \sigma_x''$ ,  $6 - \sigma_y'$ .



# Transformation Matrices

$$M(E) = \begin{pmatrix} 1 & 0 \\ 0 & 1 \end{pmatrix}, \quad M(C_6) = \begin{pmatrix} 0 & -1 \\ 1 & 1 \end{pmatrix}, \quad M(C_{-6}) = \begin{pmatrix} 1 & 1 \\ -1 & 0 \end{pmatrix},$$

$$M(C_3) = \begin{pmatrix} -1 & -1 \\ 1 & 0 \end{pmatrix}, \quad M(C_{-3}) = \begin{pmatrix} 0 & 1 \\ -1 & -1 \end{pmatrix}, \quad M(C_2) = \begin{pmatrix} -1 & 0 \\ 0 & -1 \end{pmatrix},$$

$$M(\sigma_x) = \begin{pmatrix} -1 & 0 \\ 1 & 1 \end{pmatrix}, \quad M(\sigma'_x) = \begin{pmatrix} 0 & -1 \\ -1 & 0 \end{pmatrix}, \quad M(\sigma''_x) = \begin{pmatrix} 1 & 1 \\ 0 & -1 \end{pmatrix},$$

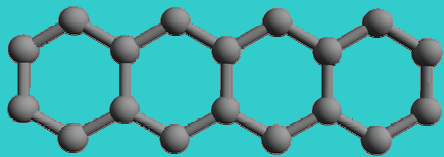
$$M(\sigma_y) = \begin{pmatrix} 1 & 0 \\ -1 & -1 \end{pmatrix}, \quad M(\sigma''_y) = \begin{pmatrix} -1 & -1 \\ 0 & 1 \end{pmatrix}, \quad M(\sigma'_y) = \begin{pmatrix} 0 & 1 \\ 1 & 0 \end{pmatrix}.$$

# Number of Polyhexes & Cost

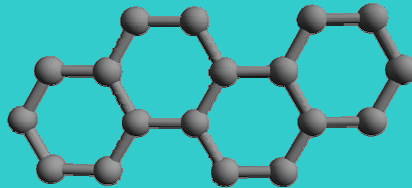
$h$	# PHs	Time (s)	$h$	# PHs	Time (s)
1	1	< 1	7	333	1
2	1	< 1	8	1448	6
3	3	< 1	9	6572	29
4	7	< 1	10	30490	167
5	22	< 1	11	143552	746
6	82	< 1	12	683101	6527



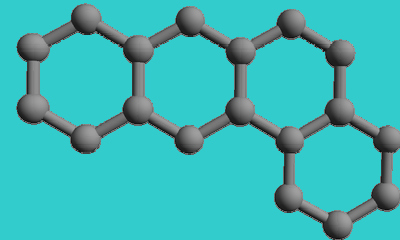
# Polyhexes with $h = 4$



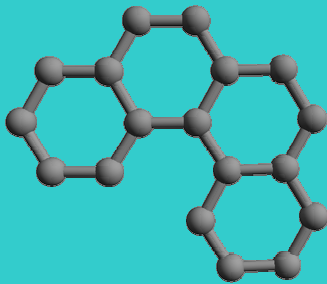
$C(1,2,3,4)$



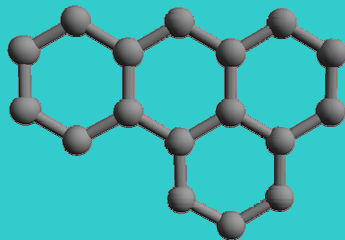
$C(3,4,5,6)$



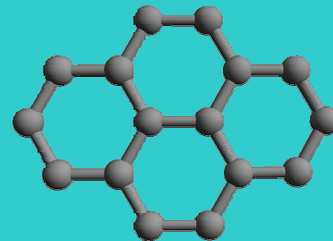
$C(1,2,3,7)$



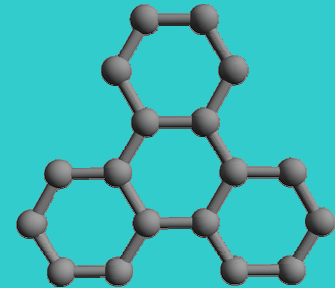
$C(2,3,5,7)$



$C(1,2,3,6)$



$C(2,3,5,6)$



$C(2,6,7,9)$

# Geometries & Electronic Transitions

- “Model geometry” – H-C-C angles:  $2\pi/3$ ; C-C bond: 1.397 Å; C-H bond: 1.084 Å.
- For geometry optimization, the Austin Model 1 (AM1) Hamiltonian is used.
- For electronic excitation energies, the Zerner’s Intermediate Neglect of Diatomic Differential Overlap (ZINDO) method is used.
- The transition energies calculated at the model geometries are found to be better than those calculated at the geometries optimized at the AM1 level of theory.



# Calibration of the Database

- The current database contains all PAHs up to  $h = 10$  ( $\sim 40,000$  PAH molecules).
- Sample space: 10 closed-shell neutral PAHs (18 rotational constants and 15 transition energies).
- Rotational constants: maximum relative error is 0.76% and the standard deviation is **0.39%**.
- Electronic transitions energies: the mean is  $-1059 \text{ cm}^{-1}$  ( $-0.13 \text{ eV}$ ); the largest deviation is  $-2757 \text{ cm}^{-1}$  ( $-0.34 \text{ eV}$ ); standard deviation is  **$1322 \text{ cm}^{-1}$  ( $0.16 \text{ eV}$ )**.

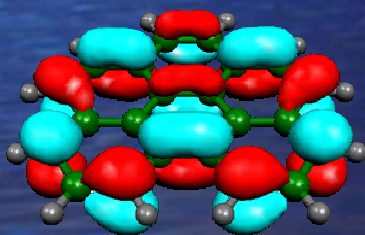
# Where the Database Was Built





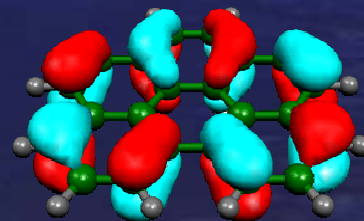
# Electronic Properties of PAHs

- The rings are in the same plane in the electronic ground state (Cs).
- Conjugated  $\pi$  ( $A''$ ) systems.
- Dominated by the  $\pi \rightarrow \pi^*$  and  $\sigma \rightarrow \sigma^*$  transitions.
- Close shell: electronic transitions are in the UV and move to longer wavelength as the size grows.
- Open shell: electronic transitions are in the visible and near IR.



LUMO

Benzoperylene

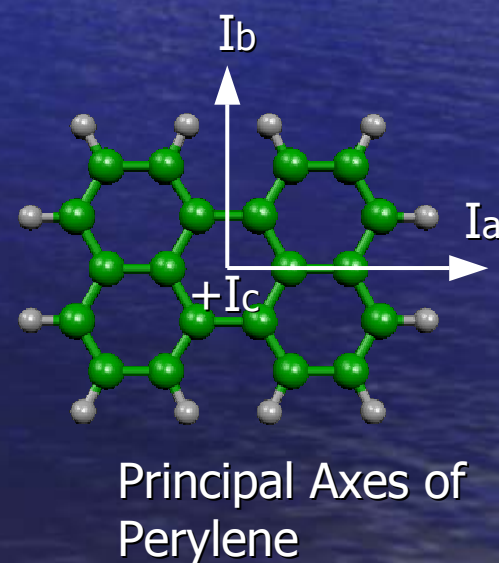
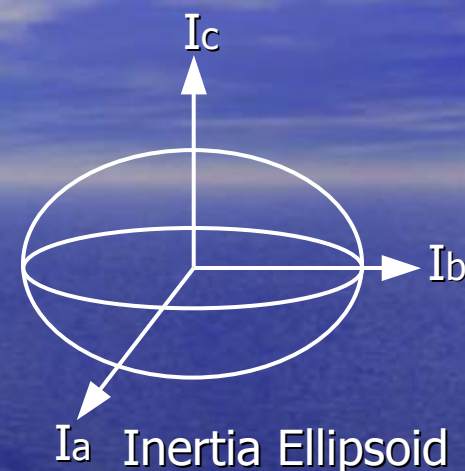


HOMO

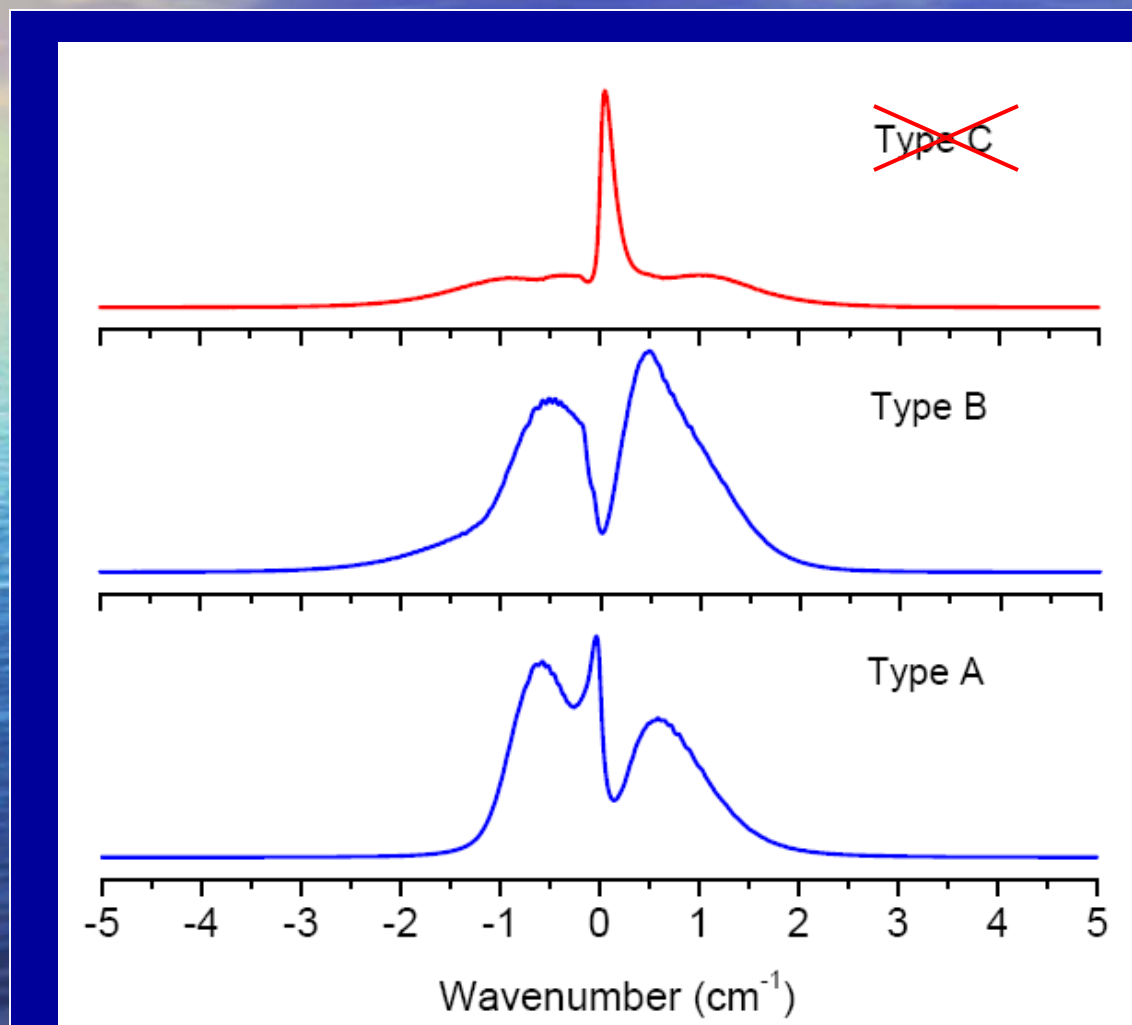


# Molecular Symmetry

- In general, PAHs are asymmetric tops.  $I_c$  is perpendicular to the molecular plane and  $I_a$  and  $I_b$  lie in the molecular plane.
- Two independent rotational constants since  $I_c = I_a + I_b$ .
- If there exists a  $C_3$  axis, the molecule is a oblate.
- Type-A and type-B band profiles.

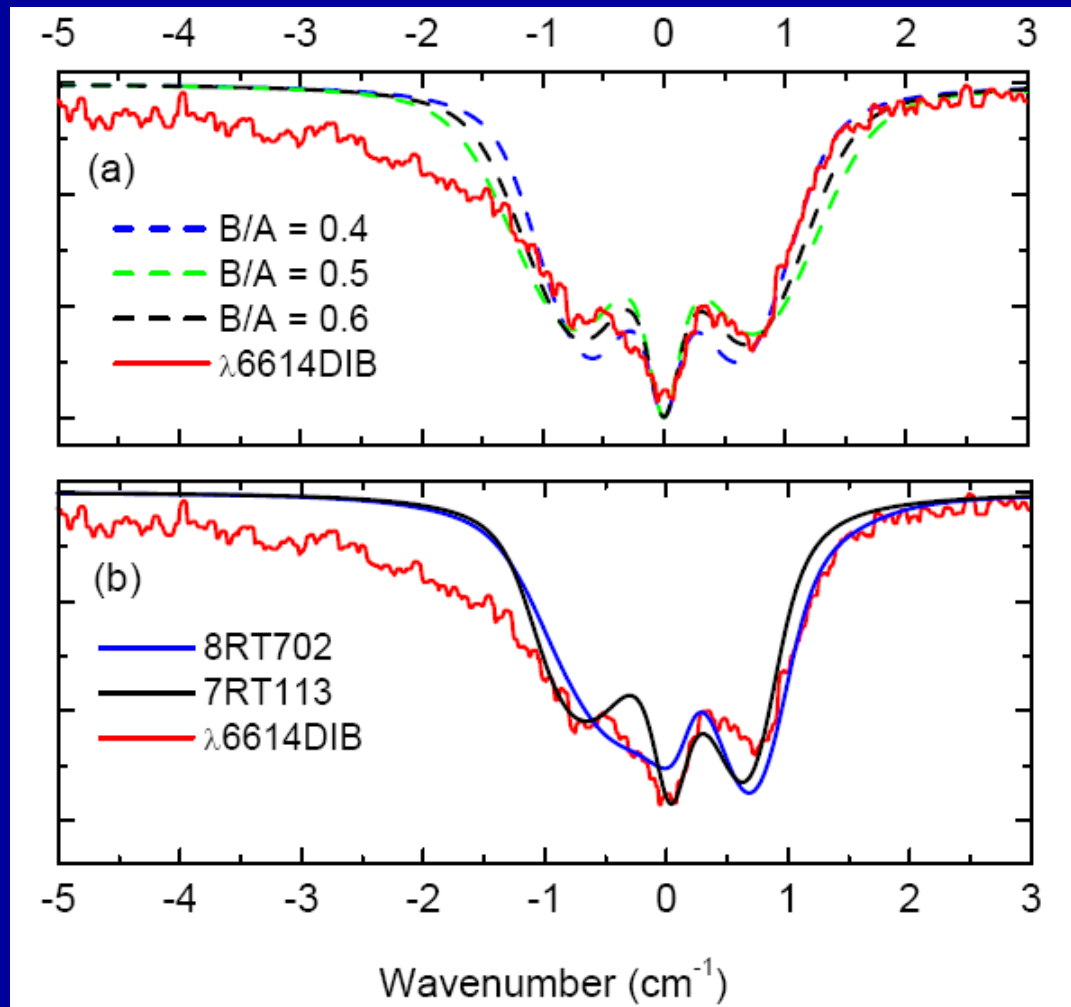


# Band Profiles – Planar PAHs



Simulated vibronic band profiles using the rotational constants of the  $S_0$  and  $S_1$  electronic states of perylene [Tan & Salama, J.C.P. 122, 084318, (2005)]. The spectra are convoluted with a Lorentzian function with  $\gamma = 0.05 \text{ cm}^{-1}$ .  $T_{\text{rot}} = 20 \text{ K}$ .

# Deciphering The $\lambda 6614$ DIB



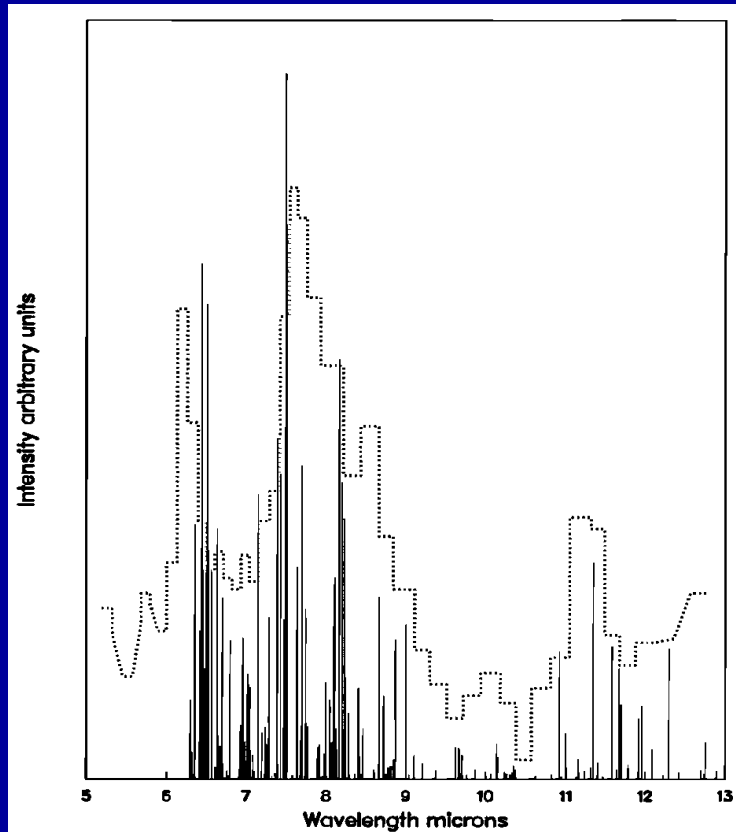
(a) Comparison of the  $\lambda 6614$  DIB (HD149757, Galazutdinov et al 2002) and simulated type-A profiles of planar PAHs,  $T_{\text{rot}} = 54$  K,  $A'' = 0.01$   $\text{cm}^{-1}$ ,  $\gamma = 0.4$   $\text{cm}^{-1}$ . In the three simulations,  $A' = A'' = A$ ,  $B' = B'' = B$ . B/A ratios are shown on the figure. (b). Comparison of the  $\lambda 6614$  DIB and simulated (TDDFT) profiles of 7RT113 and 8RT702.



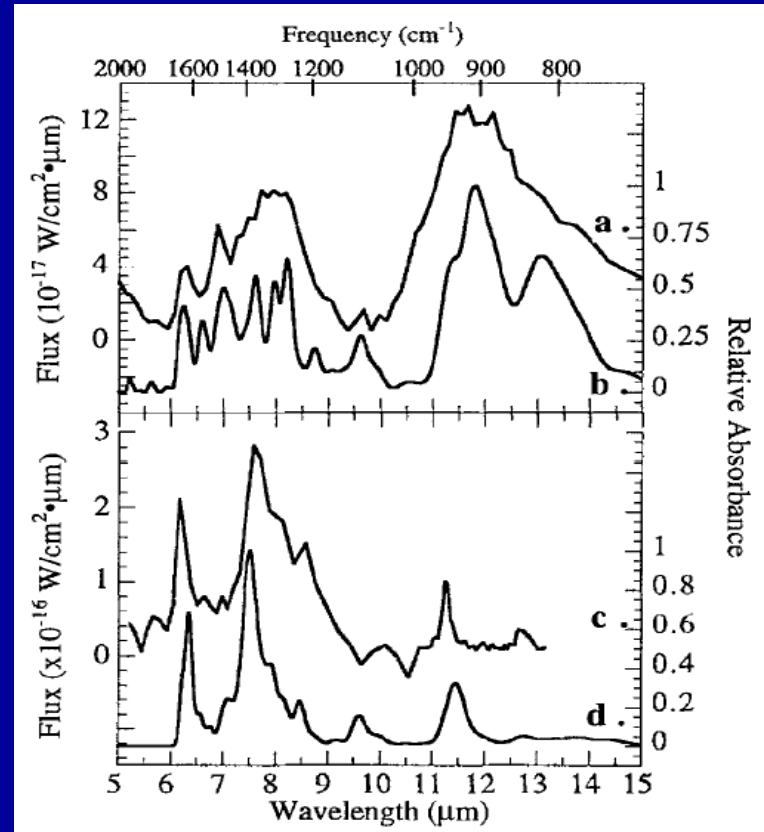
# Conclusions

- It is possible to build a “complete” electronic database of PAHs.
- Survey of all possible PAHs in this database is a very promising method to solve the PAH-DIB problem.
- For the first time, two closed-shell PAH cations are found to meet all constraints put on the  $\lambda 6614$ DIB (wavelength, intensity, band profile, and ionization potential).

# The Infrared Emission Bands



Dotted: IR spectrum of the Orion Nebula. Solid: theoretical spectrum of eight PAH cations. Langhoff, J. Phys. Chem., 100, 2819 (1996).



(a) & (c): Spectra of IRAS 22272+5435 & the Orion Nebula. (b) & (d) Spectra of mixed neutral and ionized PAHs. Allamandola et al, Ap.J. 511, L115 (1999).